



Glantreo Ltd Physical Chemistry Analytical techniques

<b>Technique</b>	<b>Principle Applications</b>	<b>Description</b>	<b>Sample Requirements</b>	<b>Comments/Limitations</b>
<b>Scanning Electron Microscope (SEM)</b>	Micron to sub-micron sample morphology and topography information. Equipped with EDS (elemental analysis) capability.	Secondary and backscattered electrons are used to generate 3D-like, high-resolution images.	Non-volatile solid sample	Magnification 50X to 200,000X. 5 nm resolution.
<b>Energy Dispersive Spectroscopy (EDX or EDS)</b>	Spatially resolved near surface (depth of a few microns) elemental analysis for detection of elements B to U. Point, line scan and mapping capabilities.	Sample is irradiated with x-rays generated in an electron microscope. Characteristic x-rays are measured based on the energy of the emitted x-rays.	Solid, non-volatile sample.	Part per thousand detection limits. Semi quantitative. 250 nm spatial resolution.
<b>Electron Spectroscopy for Chemical Analysis (XPS)</b>	Surface elemental and bonding information including element identification, depth profiling and element mapping.	Sample is irradiated with x-rays and the emitted photoelectrons are measured as a function of binding energy.	Sample must be a vacuum stable, clean solid.	Part per thousand detection limits. 30 micron lateral resolution; 30-50 Angstrom probe depth resolution.

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<b>Transmission Electron Microscope (TEM)</b>	Provision of fine, microstructure information including filler/phase dispersion and crystallinity.	High energy electrons are transmitted through a thin sample to obtain highresolution images.	Most solids and liquids. Cryo microtome and cryo plunge sample preparation.	Extensive sample preparation may be required. ~2 Angstrom resolution.
<b>X-ray Diffraction (XRD)</b>	Identification of crystalline compounds, crystalline size and percent crystallinity.	The angle and intensity of x-rays diffracted from a material are measured.	Crystalline solids, partially crystalline polymers and crystalline containing liquids.	Generally nondestructive technique.
<b>Particle Size Analysis</b>	Provision of Particle Size of sample	Laser light scattering is utilised to measure the particle size of the sample	Dry powders and liquid dispersions	Sample can be reclaimed
<b>Microanalysis</b>	Determination of %C, %H, &N and &S within a sample	Combustion techniques are employer to determine the percentages of certain elements within a saple	Solid samples	Destructive
<b>Surface area measurements (BET)</b>	Determination of surface area, pore volume and pore diameters within a samples	Nitrogen adsorption is utilised to determine the physiochemical properties of a samples	Solid/powder samples	Non destructive
<b>Differential Scanning Calorimetry (DSC)</b>	Determining melting points/phase inversion temperature/impurities etc	DSC directly measures heat changes that occur in sample during controlled increase or	Solid and liquid samples	May be particular useful with polymer species/destructive



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		decrease in temperature,		
<b>Thermal Gravimetric Analysis (TGA)</b>	Determines % water in a polymer sample	A sample is heated and its weight/mass is constantly monitored	Solids	Destructive
<b>Viscosity Measurements (via Brookfield)</b>	Determine the viscosity within a liquid/viscous solution	Brookfield rotational viscometers measure viscosity by sensing the torque required to rotate a spindle at constant speed while immersed in fluid	Liquids	Non destructive
<b>Inductively Coupled Plasma (ICP)</b>	Determination of ppm/ppb levels of elements within a sample	ICP is a type of emission spectroscopy that uses the inductively coupled plasma to produce excited atoms and ions that emit electromagnetic radiation at wavelengths characteristic of a particular element.	Liquids or solids that are easily digestible	Can generally determine to a ppb level any element within a liquid or solid sample. Solid sample generally need digestion